Chemistry teaching schedule - Form 5

Topics covered

Term	1 st term*	2 nd term	
Content	Laboratory safety & regulation (1 period)	5. Fossil fuels and carbon compounds (20 periods)	
	6. Microscopic world 2(20 periods)	11. Chemistry of carbon compounds (25 periods)	
	8. Chemical reactions and energy (15 periods)	12. Patterns in the chemical world (30 periods)	
	9. Rate of reaction (15 periods)		
	10. Chemical equilibrium (15 periods)		

^{* 61.} Detecting the presence of chemical species (part 15) will be covered on the first term.

Teaching Schedule

1st term

Content	Activity / Experiment	
Course requirements	Fire drill	
Laboratory Safety & Regulations		
6. The Microscopic World 2	Expt:	
25. Simple molecular substances with non-octet structures and shapes of simple	- Building models with different shapes	
molecules	- Effect of electrostatic field on polar and non-polar liquid	
26. Bond polarity		
27. Intermolecular forces		
28. Structures and properties of molecular crystals		
8. Chemical reactions and energy35. Energy changes in chemical reactions36. Standard enthalpy change of combustion, neutralization and formation37. Hess's law	Expt: - Enthalpy change of reactions - Application of Hess's Law - SBA (expt)	

	Expt:		
9. Rate of reactions	- rate equation		
38. Rate of chemical reaction	- effect of catalyst / temp on rate of reactions		
39. Factors affecting rate of reaction	- catalytic effect on transition element		
40. Molar volume of gases at r.t.p.	- SBA (expt)		
10. Chemical equilibrium	Expt:		
41. Dynamic equilibrium	Demo: effect of temperature and conc.		
42. Equilibrium constant	- Measure reaction rate by volume		
43. The effect of changes in concentration and temperature on chemical	- SBA (QA)		
equilibria			
Christmas Holiday			
1 st examination			

2^{nd} term

Content	Activity / Experiment	UT				
Examination Review						
5. Fossil fuels and carbon compounds						
20. Hydrocarbons from fossil fuels						
Lunar New Year Holiday						
5. Fossil fuels and carbon compounds	SBA (QA)	UT (part 9)				
21. Consequences of using fossil fuels						
22. Homologous series, structural formulae and naming of carbon compounds						
11. Chemistry of carbon compounds						
44. Introduction to selected homologous series						
45. Isomerism						
5. Fossil fuels and carbon compounds						
23. Alkane and alkenes						
Easter Holiday						

		eneminary remaining semestrate (2018 2010) / 1.8			
5. Fossil fuels and carbon compounds	SBA (VA)	UT (part 13)			
24. Addition polymers		UT (part 10)			
11. Chemistry of carbon compounds					
46. Typical reactions of various functional groups					
47. Inter-conversions of carbon compounds					
48. Important organic substances					
12. Patterns in the chemical world	SBA (VA)	UT (part 11)			
49. Periodic variation in physical properties of the elements Li to Ar					
50. Bonding, stoichiometric composition and acid-base properties of					
oxides of the elements Na to Cl					
51. General properties of transition metals					
(paper 1 review, 4 periods)					
Yearly examination					

Summer revision:

13. Industrial chemistry (Elective) 50. Importance of industrial processes + 51. Rate equation